"Tecnologías de captura, almacenamiento y usos del CO₂ Soluciones para afrontar el cambio climático" Curso de verano 1 a 5 de julio 2019 Campus de Móstoles. Aula CEI "Energía Inteligente" - URJC

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SUN to LIQUID

Fuels from concentrated sunlight

Decarbonizing Transport: European Targets

- Policy related frameworks (Renewable Energy Directive, RED 2009)
 - D Binding renewable energy targets by 2020 (10% share in transport)*
 - O Binding climate relevant targets -20% emissions by 2020, -80% emissions by 2050
 - Industry related frameworks (ACARE targets, ATAG targets)
 - Advisory Council for Aviation Research & Innovation in Europe
 - O ACARE Vision 2020 → SRA 2002-2020 (no sustainable fuel targets)
 - ACARE Flightpath 2050 \rightarrow SRIA 2020-2050 (with sustainable fuel targets)
 - Technologies and operations for **75%** CO₂ emission reduction (rel. to **2000, Aircraft**)
 - Air Transport Action Grp. (ATAG) 50% CO₂ emission reduction (rel. to 2005, Global Fleet)

• Policy/industry joint initiatives

• European Advanced Biofuels Flightpath \rightarrow 2 Mio t sustainable biofuel p.a. by 2020

* Fuel suppliers are also required to reduce the greenhouse gas intensity of the EU fuel mix by 6% by 2020 in comparison to 2010. Increased renewable energy target for 2030 from 27% to 32% (all forms of energy use including transport fuels)







The Aviation Target: Reduce CO₂ by 50% by 2050 (rel. to 2005)



 Future GHG emissions are a function of three main factors: (3) zero carbon energy



$$\dot{M}_{\rm CO_2eq}(2050) = \dot{M}_{\rm CO_2eq}(2005) \times \begin{bmatrix} {\rm GROWTH} \\ {\rm in} \\ {\rm RPK} \end{bmatrix} \times \begin{bmatrix} {\rm GAIN} \\ {\rm in} \\ {\rm EFFICIENCY} \end{bmatrix}^{-1} \times \left(1 - \begin{bmatrix} {\rm FRACTION \ of \ 2050's} \\ {\rm "ZERO-CARBON"} \\ {\rm ENERGY} \end{bmatrix} \right)$$

Key Technology: Ceria-based Redox Cycles





Source: Marxer et al, Solar thermochemical splitting of CO_2 into separate streams of CO and O_2 with high selectivity, stability, conversion, and efficiency, Energy Environ. Sci., 2017,10, 1142-1149; see also: <u>http://www.solar-jet.aero/page/media-centre/scientific-publications.php</u>

SUN-to-LIQUID Approach to Solar Fuel



Ambition: unlimited supply of renewable synthetic hydrocarbon fuels



Some process steps already proven on an industrial scale

Lowest technology readiness level for thermochemical conversion and CO₂ capture

SUN-to-LIQUID Ambition



- Move from laboratory to field environment
- Demonstration of complete fuel production cycle in a relevant environment
- Increase TRL from 3 to 5
- Scale-up of thermal power input from 4 kW to 50 kW
- Optimization of reactor geometry and material structure to increase efficiency from 2% to 5-10%
- On-site conversion of produced syngas to hydrocarbons



Key Objectives: Solar Concentration

- Key Objectives: Scale-up and experimental demonstration of the complete StL process chain to liquid hydrocarbon fuels from H₂O, CO₂ and solar energy at a pre-commercial scale H₀ Co₂
- O High flux solar concentrating system
 - 50 kW radiative power, flux of 2500 kW/m² over a 16-cm diameter aperture
 - Field of 169 heliostats, 3 m² each, with 20 – 30 m focal length









Key Objectives: 50 kW Solar Reactor



- O High flux solar concentrating system
 - 50 kW radiative power, flux of 2500 kW/m² over a 16-cm diameter aperture
 - Field of 169 heliostats, 3 m² each, with 20 – 30 m focal length

50 kW solar thermochemical reactor

 Producing syngas via ceria-based thermochemical redox cycle

Dual-Scale Ceria RPC Structures inside Reaction Cavity







Key Objectives: Gas-to-Liquid Plant

- Key Objectives: Scale-up and experimental demonstration of the complete StL process chain to liquid hydrocarbon fuels from H_2O , CO_2 and solar energy at a pre-commercial scale
- High flux solar concentrating system
 - 50 kW radiative power, flux of 2500 kW/m² over a 16-cm diameter aperture
 - Field of 169 heliostats, 3 m² each, with 20 – 30 m focal length

50 kW solar thermochemical reactor

- Producing syngas via ceria-based thermochemical redox cycle
- Gas-to-liquid system
 - Compression and storage of syngas
 - Micro Fischer-Tropsch unit converts syngas to liquid hydrocarbon fuels



CO₂/H₂O

capture/storage

Solar

concentration

| | HYPEAR |
|--|--------|
| | |



Thermo-

chemistry

Gas

storage



Fischer-

Tropsch

Com-

bustion

STEPS TO SCALING-UP SOLAR CHEMISTRY





50-300 kW

Optimization of solar concentration systems

 Combined efficiency: Receiver + Power block

$$\eta_{tot_rec} = \eta_{rec} * \eta_{Carnot}$$

• Stagnation temperature:

$$\frac{d\eta_{tot_rec}}{dT} = 0 \quad \longrightarrow \quad$$



Solar Chemical Plant Overview







Cornfield layout





Design and construction of a 50 kW high-flux solar concentration system



Heliostat with facet of 60 m radius (August 2016)



Design and construction of a 50 kW high-flux solar concentration system







Construction completed by end January 2017 (M13)



Testing and characterization of high-flux solar concentration system



• Highest measured power ever achieved: 71.5 kW onto 16-cm aperture (new calorimeter on site), solar noon on 28th June 2018, for a DNI of 800 W/m²



Control Room









Representative Solar Redox Cycle

| reduction | oxidation | $ \begin{aligned} I_{reduction, end} & 1 \\ T_{oxidation, start} & 2 \\ \dot{V}_{H_2O} & 71 \\ \dot{V}_{CO_2} & 10 \end{aligned} $ | 500 °C 900 °C .4 L/mir 0 L/min |
|--|--|--|---|
| 14 12 $(1- \operatorname{uim} 1)$ 8 | p 1500 T H_2 1400 1300 1200 | ressure (bar) 1.0 1.0 1.0 1.0 1.0 1.0 1.0 1.0 | |
| $\begin{array}{c} 6 \\ - \\ 0 \\ - \\$ | CO - 1100 - 1000 - 900 800 | Beactor 100.001 IE-4 | |
| 0 10 Time | 20 30 40 e (min) | | 2 |

Experimental Conditions

Specific test validation MS2



















| Species | Concentration | Flow (SLM) |
|-----------------|---------------|------------|
| H ₂ | 50% | 3.13 |
| CO | 15% | 0.94 |
| CO ₂ | 25% | 1.56 |
| Ar | 10% | 0.63 |
| total | 100% | 6.25 |

Design of GTL plant (FT reactor)

• Fischer Tropsch design

- Annular reactor design with oil heating and cooling
- Operating conditions for synthetic S2L syngas: 20 bar, 210°C
- Production of Liquids and wax









GtL plant: preparation





GtL plant: Fischer-Tropsch reactor





Perspectives: Resource Efficiency



• Area required for 100% substitution of European jet fuel demand





European agricultural area (2005)²: 250 Mha

60 Mha HEFA (rapeseed)³ 20 Mha BTL (woody biomass)³ 1,7 Mha STL (DNI 2000 kWh/m²)

 ¹ EIA (2008), International Energy Annual 2006
 ² FAO (2010), ResourceSTAT-Land 2005
 ³ BHL (2010), The Bauhaus Inventory of Energy Crops Mha: Million Hectare; DNI: Direct Normal Irradiation

Solar Themochemical Fuels: Water Demand



Table 2 Overall direct water footprint for the production of solar thermochemical jet fuel.

| | L L ⁻¹ jet fuel | L L ⁻¹ Naphtha | L per functional unit |
|-----------------|-------------------------------|------------------------------|--------------------------|
| Mirror cleaning | 3.62 | 3.37 | 6.54 |
| Thermochemistry | 1.66 | 1.54 | 2.99 |
| Electricity | 2.14 | 1.99 | 3.86 |
| Fotal | 7.41 | 6.90 | 13.39 |
| LOTAI | 7.41 | 0.90 | 13. |

Table 3 Overall indirect water footprint for the production of solar thermochemical jet fuel.

| Solar concentration infrastructure | L L ⁻¹ jet fuel | L L ⁻¹ Naphtha | L per functional unit |
|------------------------------------|-------------------------------|------------------------------|--------------------------|
| Heliostats | 7.10 | 6.61 | 12.83 |
| Tower | 0.95 | 0.89 | 1.72 |
| Thermochemistry | | | |
| Ceria | 31.5 | 29.4 | 56.9 |
| Alumina | 0.0031 | 0.0029 | 0.0055 |
| Stee1 | 0.043 | 0.040 | 0.078 |
| Glass | 0.011 | 0.010 | 0.020 |
| CSP infrastructure | 2.70 | 2.52 | 4.88 |
| Fischer-Tropsch infrastructure | | | |
| Steel | 0.0044 | 0.0041 | 0.0080 |
| Concrete | 0.00062 | 0.00057 | 0.0011 |
| Total | 42.4 | 39.4 | 76.5 |



• For comparison: Water footprint of biofuels from 500 to 18 000 L L-1 jet fuel

equivalent

Source: C. Falter, R. Pitz-Paal, Water footprint and land requirement of solar thermochemical jet fuel production, Environmental Science and Technology 2017, doi: 10.1021/acs.est.7b02633, accepted for publication.

Perspectives: Resource Efficiency



- Use of biofuels is controversial
 - Biofuels are available (TRL 9) and approved for civil aviation (e.g. FT-SPK, HEFA, AtJ)
 - O Controversial environmental performance
 - Relatively low area specific yield
 - O High water demand
 - Limited GHG reduction potential (LUC)
- Solar fuel production from H_2O and CO_2
 - Large GHG reduction potential
 - Resource efficiency: High yield, no arable land required, very low water consumption
 - Complementary production to biofuels



Data: C. Falter, Climate Impact and Economic Feasibility of Solar Thermochemical Jet Fuel Production, Environ. Sci. Technol., 2016, 50 (1)

German Environment Agency (UBA), Power-to-Liquids Potentials and Perspectives for the Future Supply of Renewable Aviation Fuel, 2016, Authors: LBST, Bauhaus Luftfahrt M. S. Wigmosta et al., National microalgae biofuel production potential and resource demand, Water Resour. Res., 47, W00H04, 2011

Area-specific range



• How much CO₂ would be required for 100% substitution with solar jet fuel?

- Stoichiometric demand to substitute current jet fuel: 850 Mt_{co2}
 - 2017 consumption 339 billion liters, specific demand about 3.14 kg_{CO2}/kg_{jet}
- O Adjusted demand: About a factor of two(GtL selectivity towards jet, fugitive losses)
- 2030 scenario: Continued growth at 5% per year
 - Average growth rate during past five years (above historic average)

| Substitution target | Stoichiometric demand | Adjusted demand |
|---------------------------------|--------------------------|-----------------------------|
| 2017: Jet fuel consumption | 850 Mt _{co2} | ca. 1 700 Mt _{co2} |
| 2030 demand (5% growth/year) | 1 600 Mt _{co2} | 3 200 Mt _{co2} |
| Current diesel for road freight | 2 000 Mt _{co2} | 4 000 Mt _{co2} |
| Current oil consumption | 14 000 Mt _{co2} | NA |

Sources: Jet fuel: IATA, *Economic Performance of the Airline Industry - 2017 mid-year report;* Road freight: IEA/OECD, *The Future of Trucks Implications for energy and the environment, 2017;* Crude Oil: *BP Statistical Review of World Energy June 2017* Assumptions: Stoichiometry for CH_2 synthesis (14 amu) from CO_2 (44 amu): $CO_2 + H_2O \rightarrow "CH_2" + 3/2 O_2$; density jet 0.8 kg/liter



Impact: Emission Reduction Potential



 Consideration of further carbon emissions along production chain can result in higher life-cycle emission compared to conventional fuel







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